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## PALLADIUM CATALYZED SELECTIVE MONOARYLATION OF 2-AMINOPYRIMIDINES AND 2-AMINOPYRAZINE WITH 1,2-DIBROMOBENZENE WITHOUT CYCLIZATION

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#### **ABSTRACT:**

A simple one-flask highly selective method for the Pd-catalyzed preparation of sterically hindered 2-(2-bromophenyl)pyrimidines.was elaborated. 2-(2-Bromophenyl)pyrimidines were isolated in 22-64% yields. Similarly was prepared 2-(2-bromophenyl)aminopyrazine.

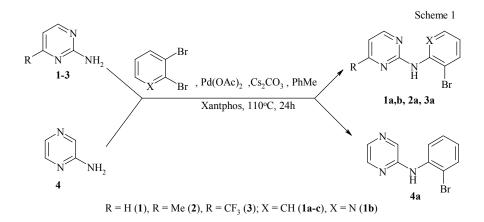
**Keywords:** palladium catalysis, 2-aminopyrimidines, 1,2-dibromobenzene, 2-(2-bromophenyl)pyrimidines

#### INTRODUCTION

Substituted 2-aminopyrimidines are of great interest as intermediates in the synthesis of organic compounds <sup>1</sup>. 2-Arylaminopyrimidines were prepared from 2-aminopyrimidines and bromobenzene or iodobenzene in the presence of CuI <sup>II</sup> or *t*-BuOK <sup>III</sup>,<sup>IV</sup>. Second group of methods of synthesis of substituted 2-arylaminopyrimidines is based on the interaction of 2chloropyrimidines with anilines in the presence of palladium<sup>I, V</sup> or cobalt<sup>VI</sup> catalysts or treatment of 2-bromopyrimidine with aniline under microwave irradiation. There are only one method for the synthesis of 2-(2-bromophenyl)aminopyrimidine based on the treatment of 2chloropyrimidine with 2-bromoaniline in high temperatures <sup>VII</sup>. Beside this, cyclization of 2pyridines (2-bromophenyl)aminopyrimidines or may form coresponding pyrimidobenzimidazoles pyridobenzimidazoles. or For example, 2-(2bromophenvl)aminopyrimidines in the presence of copper catalvst afforded pyrimidobenzimidazoles in almost quantitative yield <sup>VIII</sup>. We have elaborated novel and for the preparation sterically Pd-catalyzed method hindered 2-(2selective bromophenyl)pyrimidines. Beside this, formation of pyrimidobenzimidazoles does not occur in these cases.

#### **RESULTS AND DISCUSSION**

We have elaborated effective catalytic system for the selective preparation of 2-(2-bromophenyl)pyrimidnes 1-3a from corresponding 2-aminopyrimidines 1-3. Thus, amines 1-3 in the system 1,2-dibromobenzene /  $Pd(OAc)_2$  / Xantphos /  $Cs_2CO_3$  / PhMe afforded 2-(2-bromophenyl)pyrimidines 1-3a-c in 22-64% yield (Scheme 1). Similarly was prepared 2-(3-bromopyridin-2-yl)aminopyrimidine (1b) (yield 52%) from amine 1 and 2,3-dibromopyridine. Reaction of 2-aminopyrazine 4 with 1,2-dibromobenzene leads to 2-(2-bromophenyl)aminopyrazine (4a) in 30% yield.



## EXPERIMENTAL SECTION

<sup>1</sup>H and <sup>13</sup>C NMR spectra were registered on Varian Mercury BB 400 MHz in CDCl<sub>3</sub>. Massspectra were recorded on Alliance Waters 2695 instrument with Full scan POS NEG 16 detector. 2-Aminopyrimidines, 1,2-dibromobenzene, palladium acetate, Xantphos and cesium carbonate (all Acros) were used without purification. Melting points were detected on Boetius aparatos equipped with visual detector PHMH 05. HR-MS spectra were performed on Micromass *Q-TOF* Micro quadrupole-time of fight high resolution mass spectrometer. Leucine enkephalin was used as internal lock mass for accurate mass calculation.

General procedure for the synthesis of 2-(2-bromophenyl)aminopyrimidines 1-3a and 2-(3-Bromopyridin-2-yl)aminopyrimidine (1b). Solid  $Cs_2CO_3$  (0.98 g, 3 mmol) was added to the solution of 2-aminopyrimidine 1-3 (1 mmol) and 1,2-dibromobenzene or 2,3dibromopyridine (1 mmol), Pd(OAc)<sub>2</sub> (11 mg, 0.05 mmol), Xantphos (29 mg, 0.1 mmol) in dry toluene (2 ml) in a glass reactor (50 ml) under argon. The reaction mixture was stirred at 110°C (TLC-control) for 24 h under argon. The solvent was removed under reduced pressure and crude residue was chromatographed on silica using  $CH_2Cl_2$  : ethyl acetate EtOH (10:3:0.4) as eluent.

**2-(2-Bromophenyl)pyrimidine (1a)** <sup>VII</sup>. Yield 62 %. M.p. 44-46°C. GH-MS: 250.0 (M<sup>+</sup>, 11), 170.1 (100), 85.1 (12). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 6.70 (1H, t, J = 4.8 Hz, H-5 in pyrimidine); 6.91 (1H, t, J = 7.2 Hz, C<sub>6</sub>H<sub>4</sub>); 7.33 (1H, t, J = 7.2 Hz, C<sub>6</sub>H<sub>4</sub>); 7.56 (1H, d, J = 7.2 Hz, C<sub>6</sub>H<sub>4</sub>); 7.60 (1H, bs, NH); 8.46 (2H, , J = 4.8 Hz, H-4 and H-6 in pyrimidine); 8.51 (1H, d, J = 7.6 Hz, C<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR (100.58 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 113.31; 113.35; 120.65; 123.36; 128.02; 132.41; 137.15; 157.99; 159.76. Found, *m/z* (EI): 249.9968 [M]<sup>+</sup>. C<sub>10</sub>H<sub>8</sub>BrN<sub>3</sub>. Calculated, *m/z*: 249.9980.

**4-Methyl-2-(2-bromophenyl)pyrimidine (2a).** Yield 64 %. M.p. 55-57°C. GC-MS, 264.0 (M<sup>+</sup>, 8), 184.1 (100), 92 (10). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 2.44 (3H, s, Me); 6.65 (1H, d, J = 5.2 Hz, H-5 in pyrimidine); 6.88 (1H, t, J = 7.2 Hz, C<sub>6</sub>H<sub>4</sub>); 7.32 (1H, t, J = 7.2 Hz, C<sub>6</sub>H<sub>4</sub>); 7.55 (1H, d, J = 7.2 Hz, C<sub>6</sub>H<sub>4</sub>); 7.56 (1H, bs, NH); 8.46 (1H, J = 5.2 Hz, H-6 in pyrimidine); 8.56 (1H, d, J = 7.6 Hz, C<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR (100.58 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 24.15; 112.96; 113.07; 120.44; 123,00; 127.98; 132.34; 137.40; 157.40; 159.55; 168.25. Found, *m/z* (EI): 264.0151 [M]<sup>+</sup>. C<sub>11</sub>H<sub>11</sub>BrN<sub>3</sub>. Calculated, *m/z* 264.0136.

**4-Trifluoromethyl-2-(2-bromophenyl)pyrimidine (3a)**. Yield 22 %. M.p. 94-96°C. GC-MS, 318.0 (M<sup>+</sup>, 10), 238.1 (100), 169.1 (9), 119.0 (10). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 6.96 (1H, t, J = 7.2 Hz, C<sub>6</sub>H<sub>4</sub>); 7.08 (1H, d, J = 5.2 Hz, H-5 in pyrimidine); 7.36 (1H, t, J = 7.2 Hz, C<sub>6</sub>H<sub>4</sub>); 7.59 (1H, d, J = 7.2 Hz, C<sub>6</sub>H<sub>4</sub>); 7.82 (1H, bs, NH); 8.46 (1H, d, J = 7.6 Hz, C<sub>6</sub>H<sub>4</sub>); 8.67 (1H, d, J = 5.2 Hz, H-6 in pyrimidine). <sup>13</sup>C NMR (100.58 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 108.35; 113.62; 118.96; 120.90; 121.70; 124.18; 128.19; 132.50; 136.34; 156.27; 160.52. Found, *m*/*z* (EI): 317.9870 [M]<sup>+</sup>. C<sub>11</sub>H<sub>7</sub>BrF<sub>3</sub>N<sub>3</sub>. Calculated, *m*/*z* 317.9854.

**2-(3-Bromopyridin-2-yl)aminopyrimidine (1b).** Yield 52 %. M.p. 154-156°C. GC-MS, 251.0 (M<sup>+</sup>, 5), 171.1 (100), 144.1 (7), 79.1 (8). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 6.85 (1H, t, J = 4.8 Hz, H-5 in pyrimidine); 6.88 (1H, m, 5-H in pyridine); 7.85 (1H, d, J = 8.0 Hz, H-4 in pyridine); 7.92 (1H, bs, NH); 8.43 (1H, d, J = 4.0 Hz, H-6 in pyridine); 8.57 (2H, d, J = 4.8 Hz, H-4 and H-6 in pyrimidine). <sup>13</sup>C NMR (100.58 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 109.71; 114.60; 119.08; 141.01; 147.21; 149.55; 158.32; 158.92. Found, *m/z* (EI): 250.9917 [M]<sup>+</sup>. C<sub>9</sub>H<sub>7</sub>BrN<sub>4</sub>. Calculated, *m/z* 250.9932.

**2-(2-Bromophenyl)aminopyrazine (4a).** Yield 30 %. M.p. 77-79°C. GC-MS, 250.0 (M<sup>+</sup>, 10), 170.1 (100), 143.0 (9), 85.0 (11), 68.1 (10). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 6.94 (1H, t, J = 7.6 Hz, C<sub>6</sub>H<sub>4</sub>); 6.95 (1H, bs, NH); 7.32 (1H, t, J = 7.6 Hz, C<sub>6</sub>H<sub>4</sub>); 7.58 (1H, d, J = 7.6 Hz, C<sub>6</sub>H<sub>4</sub>), 8.04 and 8.15 (2H, both d, J = 2.8 Hz, H-5 and H-6 in pyrazine); 8.27 (1H, d, J = 1.2 Hz, H-3 in pyrazine); 8.50 (d, J = 7.6 Hz, C<sub>6</sub>H<sub>4</sub>). <sup>13</sup>C NMR (100.58 MHz, CDCl<sub>3</sub>)  $\delta$  (ppm): 114.23; 120.30; 123.92; 128.24; 132.80; 134.08; 135.76; 137.18; 141.66. Found, m/z (EI): 250.0006 [M]<sup>+</sup>. C<sub>10</sub>H<sub>8</sub>BrN<sub>3</sub>. Calculated, m/z 249.9980.

# REFERENCES

- I. D.S. Ermolat'ev, E.V. Van der Eycken, J. Org. Chem. 73, 6691 (2008).
- II. Y. Liu, Y. Bai, J. Zhang, Y. Li, J. Jiao, X. Qi, Eur. J. Org. Chem. 6084 (2007).
- III. Y. Heo, D. Hyun, M.R. Kumar, S. Leo, H.M. Jung, S. Lee, Tetrahedron Lett. 53, 6657 (2012).
- IV. A.J. Kim, H.J. Lee, J.C. Park, H. Kang, H. Yang, H. Song, K.H. Park, Molecules 14, 5169 (2009).
- V. X. Hao, J. Yuan, G.-A. Yu, M.-Q. Qiu, N.-F. She, Y. Sun, C. Zhao, S.-L. Mao, J. Yin, S.-H. Liu, J. Organomet. Chem. **706-707**, 99 (2012).
- VI. L.B. Delvos, J.-M. Begouin, C. Gosmini, Synlett 2325 (2011).
- VII. W.A.W. Stolle, A.T.M. Marcelis, A. Koetsier, H.C. van der Plas, Tetrahedron 45, 6511 (1989).
- VIII. K. Liubchak, K. Nazarenko, A. Tolmachev, Tetrahedron, 68, 2993 (2012).

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